Divergent and Selective Functionalization of 2-Formylpyrrole and its Application in the Total Synthesis of the Aglycone Alkaloid Pyrrolemarumine

Eder I. Martínez-Mora,[†] Miguel A. Caracas,[†] Carlos H. Escalante, Damian A. Madrigal, Héctor Quiroz-Florentino, Francisco Delgado, and Joaquín Tamariz*

Departamento de Química Orgánica, Escuela Nacional de Ciencias Biológicas,

Instituto Politécnico Nacional, Prolongación de Carpio y Plan de Ayala S/N. 11340, México, D.F., Mexico

Fax: +5255-5729-6300/46211; E-mail: jtamarizm@gmail.com; jtamariz@woodward.encb.ipn.mx;

[†] Equivalent contributions

* Corresponding author

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Abstract. Diverse 1,2- and 1,2,5-substituted pyrroles were efficiently prepared through a regioselective functionalization of 2-formylpyrrole (**5a**). This methodology was applied for the first total synthesis of pyrrolemarumine (**4b**), the aglycone of the corresponding natural pyrrole alkaloid 4"-O- α -L-rhamnopyranoside. The synthesis of **4b** was achieved starting from **5a** through a seven-step process in 28% overall yield.

Key words: 2-formylpyrrole; Vilsmeier-Haack formylation; pyrrole alkaloids; pyrrolemarumine.

Introduction

Pyrroles are ubiquitous five-membered heterocycles forming part of the structure of a large number of natural products [1] and pharmacologically active compounds [2]. Alkaloids isolated from diverse natural sources, such as higher order plants [3] and marine species [4] display a substituted single pyrrolic ring or pyrrole-fused polycyclic or heterocyclic scaffolds, which are distinguished by their strong antibiotic, anticancer, antifeedant and antiviral activity.

Pyrrole-2-aldehyde derivatives represent a unique variety of alkaloid natural products isolated from fungi, microorganisms, plants, and edible fruits, among other natural sources. For example, jiangrines A-E (1a-e) and pyrrolezanthine (1f), which exhibit anti-inflammatory activity, were isolated from the fermentation broth of Jiangella gansuensis (Fig. 1) [5]. Fusarine (1g) is a naturally occurring 2-acyl pyrrole isolated from the culture broth of Fusarium incarnatum (HKI0504), an endophytic fungus of the mangrove plant Aegiceras corniculatum [6]. Makomotines **2a-c** were isolated from an edible gall called Makomotake (Zizania latifolia infected with Ustilago esculenta) found in Japan, China and other Asian countries [7]. Pyrrole alkaloids 2d-g have been isolated from an extract of the fruits of Lycium chinense Miller (Solanaceae), which is used as a traditional tonic medicine for treating liver and kidney failures [8]. From the seeds of watermelon (Citrullus lanatus (Thunb.)), **Resumen.** Se prepararon eficientemente diversos pirroles 1,2- y 1,2,5-subtituidos a través de la funcionalización regioselectiva del 2-formilpirrol (**5a**). Esta metodología se aplicó en el desarrollo de la primera síntesis total de la pirrolemarumina (**4b**), que es la aglicona del alcaloide pirrólico natural correspondiente 4"-O- α -L-ramnopiranósido, la cual se realizó partiendo de **5a** a través de un proceso en siete etapas y en un rendimiento global de 28%.

Palabras clave: 2-formilpirrol; formilación de Vilsmeier-Haack; alcaloides pirrólicos; pirrolemarumina.

2-formyl pyrroles **3a-b** were isolated that exhibit modest inhibitory activity on melanogenesis [9]. The naturally occurring pyrrole alkaloid pyrrolemarumine 4"-O- α -L-rhamnopyranoside (**4a**), recently isolated from leaves of *Moringa oleifera* Lam., was hydrolyzed to yield the new aglycone pyrrolemarumine (**4b**) [10] (Fig. 2). Despite their potential biomedical properties and relatively simple structure, the synthesis for most of these unusual 1,2- and 1,2,5-substituted pyrrole alkaloids has not yet been reported [11]. Therefore, a synthetic approach to any of these compounds needs to be designed on the basis of the selective functionalization of pyrrole or 2-formylpyrrole (**5a**).

The structure and promising pharmacological profile of these compounds fits well into our ongoing research program of carrying out the transformation of simple five-membered heterocycles into fine chemicals and more complex natural products [12]. Hence, we herein investigated the reactivity of 2-formylpyrrole (**5a**) as the key starting material for the synthesis of a series of 1,2- and 1,2,5-trisubstituted pyrroles, as well as in the first total synthesis of compound **4b**.

Results and discussion

Synthesis of 1,2-Disubstituted Pyrroles

2-Formylpyrrole (5a) was used as the starting material for the divergent synthesis [13] of novel 1,2-substituted pyrrole

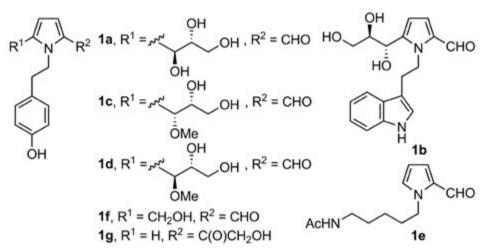


Fig. 1. Structures of jiangrines A-E (1a-e), pyrrolezanthine (1f) and fusarine (1g).

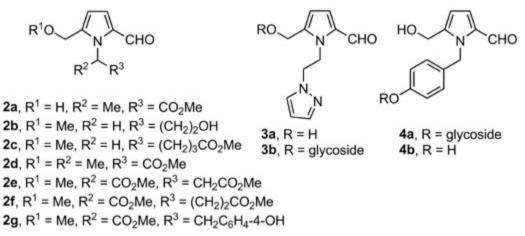


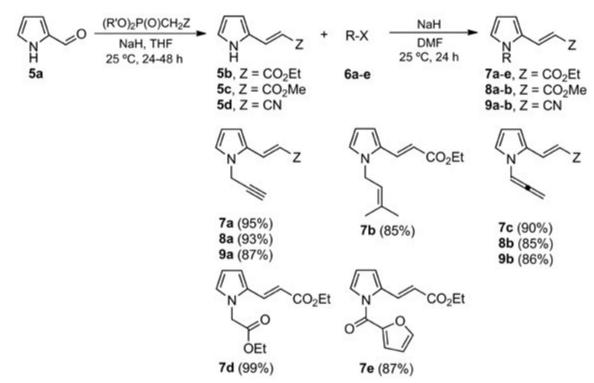
Fig. 2. Structures of makomotines 2a-c, 2-formyl pyrroles 2d-g and 3a-b, pyrrolemarumine 4"-O-a-L-rhamnopyranoside (4a) and pyrrolemarumine (4b).

derivatives of the ethyl 3-acrylates **7a-e**, methyl acrylates **8a-b** and acrylonitriles **9a-b** (Scheme 1). These vinylogous electron-deficient pyrroles were selected because they may be applied as potential precursors for the preparation of more complex and polysubstituted pyrroles as HMG-CoA reductase inhibitors [14], whose pharmacological activity is also found in our potent hypolipidemic compounds [15].

Previous studies have reported the direct *N*-alkylation by NaH-promoted deprotonation of commercially available **5a** with diverse primary alkyl halides to furnish the series of 1-substituted 2-formylpyrroles [16]. This method was also useful for the preparation of the series of new 1,2-substituted pyrroles **7-9** starting from **5b-d**, which are the Horner-Wadsworth-Emmons derivatives of **5a**. Thus, under mild reaction conditions, pyrroles **7a-e** were synthesized in high yields (Scheme 1). Pyrrole **7c** was prepared in a single-step procedure by using **5b**, propargyl bromide (**6a**) and an excess of NaH. Similarly, in the case of methyl acrylate **5c**, the reaction with **6a** provided either *N*-propargyl pyrrole **8a** or *N*-allenyl pyrrole **8b** in high yields. The latter was generated by direct isomerization of **8a** or through the cascade alkylation of **5c** with **6a** in the presence of an excess of NaH, similar to **7c**. Likewise, *N*-alkylation of 3-(pyrrol-2-yl)acrylonitrile (**5d**) led to *N*-substituted pyrroles **9a-b** in good yields, under similar mild reaction conditions.

Regioselective Synthesis of 1,2,5-Substituted Pyrroles

Evaluation of the reactivity and regioselectivity of the formylation of pyrroles **7** was exemplified by using pyrroles **7a-c**. Thus, the latter were formylated under the usual Vilsmeier-Haack conditions to give rise to the corresponding 5-formyl derivatives **10a-c** in good yields (81-88%) (Scheme 2). Interestingly, in all the substrates the C-5 formyl regioisomer was exclusively obtained and no mixtures of the three possible C-3/C-4/C-5 formyl isomers were observed. This is in agreement with previous reports for an analogous substrate [17], though there is a broad tendency to provide no selective ratios of regioisomers [17,18]. Divergent and Selective Functionalization of 2-Formylpyrrole and its Application in the Total Synthesis



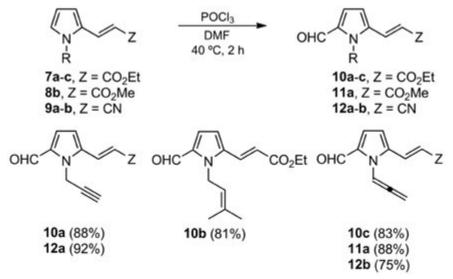
Scheme 1. Preparation of 1,2-disubtituted pyrroles 7a-e, 8a-b, and 9a-b.

A behavior similar to pyrroles **7a-c** was found when methyl acrylate **8b** and acrylonitriles **9a-b** were formylated to afford the corresponding 1,2,5-trisubstituted pyrroles **11a** and **12a-b**, respectively.

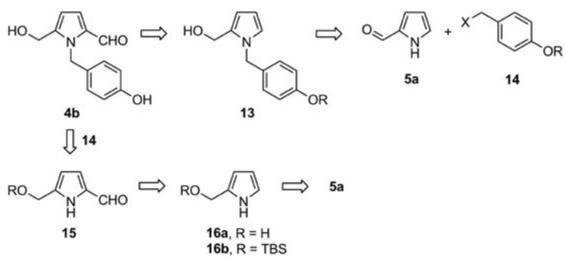
The highly regioselective preparation of these 1,2,5-trisubstituted pyrroles in good overall yields prompted us to explore the use of 2-formylpyrrole (5a) as an efficient starting material for the total synthesis of a naturally occurring 2-formyl pyrrole alkaloid, such as pyrrolemarumine (4b). The latter compound was chosen because most of the natural alkaloids illustrated in figures 1-2 display the same 2,5-functionalities in the pyrrole core.

Total Synthesis of Pyrrolemarumine (4b)

The total synthesis of pyrrolemarumine (4b), the aglycone of the natural pyrrole alkaloid 4a, was designed based on the insights gained from observing the behavior of 2-formylpyrrole



Scheme 2. Synthesis of formyl pyrroles 10a-c, 11a and 12a-b.

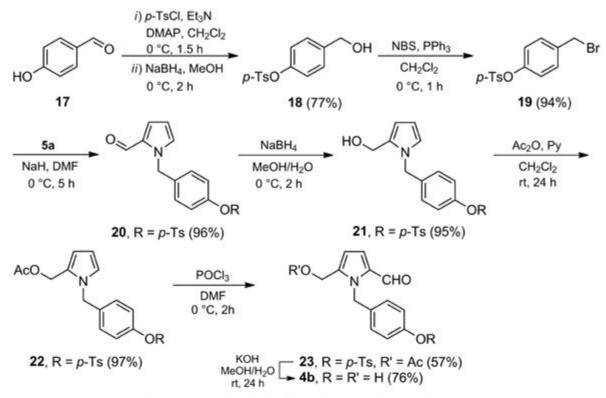


Scheme 3. Dual retrosynthesis for the preparation of the aglycone pyrrolemarumine (4b).

(5a) with the diverse reagents herein described. A dual retrosynthetic scheme was proposed (Scheme 3), which included the approach starting from the *N*-benzylation of 5a, followed by formylation of intermediate 13 to furnish the desired product 4b. The alternative more convergent pathway would be the *N*-benzylation of 2,5-disubstituted pyrrole 15, which would be previously functionalized from 5a.

Because of the obvious advantages of a convergent synthesis, the second approach was investigated first. Although pyrrole **5a** was readily reduced with sodium hydride to yield the corresponding 2-hydroxymethyl pyrrole (**16a**), this had to be protected with a TBS group to afford the silane derivative **16b** and in this way avoid decomposition. However, further degradation of this substrate under the formylation conditions led us to abandon this route and attempt the first approach.

In order to introduce the benzyl moiety into the pyrrole framework, the synthetic route followed the reaction conditions depicted in Scheme 1, whereby the benzyl bromide derivative



Scheme 4. Total synthesis of the aglycone pyrrolemarumine (4b).

19 gave 2-formylpyrrole **20** in high yield (Scheme 4). Derivative **19** was prepared from 4-hydroxybenzaldehyde (**17**) in good overall yield (72%) through a two-step reaction sequence, including intermediate **18**.

The reduction of **20** with sodium borohydride provided the corresponding alcohol **21** in high yield, followed by acetylation to yield acetate **22**. This protection was provided due to the instability shown by the hydroxyl group during the subsequent formylation step, which is similar to what occurred during the first approach. The latter reaction carried out under standard conditions brought the formyl group into the desired position of the pyrrole ring, resulting in **23** in a modest yield. Hydrolysis of the protective groups of the latter compound furnished the desired product **4b** in a 43% overall yield for the two steps, and in a 28% overall yield for the seven steps starting from **17**. The spectral data of the synthetic product **4b** was in agreement with the data reported for the aglycone natural product [10].

Conclusions

In summary, a divergent synthetic approach for the preparation of 1,2-di- and 1,2,5-tri-substituted pyrroles starting from 2-formylpyrrole (**5a**) has been achieved, including the first total synthesis of the aglycone alkaloid pyrrolemarumine (**4b**) in a high overall yield. The scope and efficiency of this approach is currently under evaluation for the synthesis of other 1,2,5-trisubstituted pyrrole alkaloids from the series of natural compounds **1-3**, and the results will be reported in due course.

Experimental Section

General: Melting points were determined with an Electrothermal capillary melting point apparatus. IR spectra were recorded on a Perkin-Elmer 2000 spectrophotometer. ¹H and ¹³C NMR spectra were recorded on Varian Mercury (300 MHz) and Varian VNMR (500 MHz) instruments, with CDCl₃ as the solvent and TMS as internal standard. Signal assignments were based on 2D NMR spectra (HMQC, HMBC). Mass spectra (MS) were recorded on Polaris Q-Trace GC Ultra (Finnigan Co.) and Hewlett-Packard 5971A spectrometers. High-resolution mass spectra (HRMS), in electron impact mode, were obtained with a Jeol JSM-GCMateII apparatus. Elemental analyses were performed on a CE-440 Exeter Analytical instrument. Analytical thin-layer chromatography was carried out using E. Merck silica gel 60 F254 coated 0.25 plates, visualized by using a longand short-wavelength UV lamp. Flash column chromatography was performed over Natland International Co. silica gel (230-400 and 230-400 mesh). All air moisture sensitive reactions were carried out under N2 using oven-dried glassware. THF was freshly distilled over sodium, as was DMF and CH₂Cl₂ over CaH₂, prior to use. MeOH was distilled over sodium. Et₃N was freshly distilled from NaOH. All other reagents were used without further purification.

Ethyl (E)-3-(1H-Pyrrol-2-yl)acrylate (5b) [19]: To a solution of 5a (0.100 g, 1.05 mmol) in anhydrous THF (2 mL) at 0 $^{\circ}$ C, NaH (60%) (0.042 g, 1.05 mmol) was added. The mixture was stirred at 0 °C under nitrogen for 30 min and triethyl phosphonoacetate (0.235 g, 1.05 mmol) was added dropwise. After stirring at room temperature for 48 h, EtOAc (30 mL) was added, the mixture washed with water $(2 \times 15 \text{ mL})$, the organic layer dried (Na₂SO₄) and the solvent removed under vacuum. The residue was purified by column chromatography over silica gel (30 g/g crude, hexane/EtOAc, 98:2) to give 5b (0.157 g, 90%) as a reddish solid. $R_f = 0.55$ (hexane/EtOAc, 7:3); mp 58-59 °C. IR (film): \overline{v} = 2981, 1683, 1622, 1544, 1446, 1366, 1268, 1181, 973, 736 cm⁻¹. ¹H NMR (300 MHz, CDCl₂): $\delta = 1.31$ (t, J = 7.1 Hz, 3H, CO₂CH₂CH₃), 4.24 (q, J = 7.1 Hz, 2H, $CO_2CH_2CH_3$), 6.09 (d, J = 15.9 Hz, 1H, H-2), 6.25-6.29 (m, 1H, H-4'), 6.53-6.58 (m, 1H, H-3'), 6.89-6.94 (m, 1H, H-5'), 7.59 (d, J = 15.9 Hz, 1H, H-3), 9.34 (br s, 1H, NH). ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 14.3$ (CO₂CH₂CH₃), 60.3 (CO₂CH₂CH₂), 110.7 (C-4'), 111.0 (C-2), 114.2 (C-3'), 122.6 (C-5'), 128.4 (C-2'), 134.6 (C-3), 168.1 (CO₂Et). MS (70 eV): $m/z = 165 (12) [M]^+, 164 (22) [M - 1]^+, 150 (5), 138 (42), 88$ (100), 86 (18), 56 (12).

Methyl (*E*)-3-(1*H*-Pyrrol-2-yl)acrylate (5c) [20]: Following the method of preparation for 5b, by using 5a (0.500 g, 5.26 mmol), NaH (60%) (0.252 g, 6.31 mmol), and trimethyl phosphonoacetate (1.140 g, 6.31 mmol) in dry THF (5 mL) and stirring at 25 °C for 24 h, 5c (0.782 g, 98%) was obtained as a colorless solid. R_f = 0.63 (hexane/EtOAc, 7:3); mp 78-79 °C. IR (KBr): \bar{v} = 3261, 2947, 1683. 1628, 1552, 1328, 1414, 1303, 1231, 1193, 1132, 1030, 966, 844, 755, 738 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 3.75 (s, 3H, CO₂CH₃), 6.12 (d, *J* = 15.9 Hz, 1H, H-2), 6.22-6.27 (m, 1H, H-4'), 6.52-6.56 (m, 1H, H-3'), 6.88-6.92 (m, 1H, H-5'), 7.59 (d, *J* = 15.9 Hz, 1H, H-3), 9.70 (br s, 1H, NH). ¹³C NMR (75.4 MHz, CDCl₃): δ = 51.4 (CO₂CH₃), 110.2 (C-2), 110.6 (C-4'), 114.5 (C-3'), 122.7 (C-5'), 128.3 (C-2'), 134.9 (C-3), 168.6 (CO₂CH₃). MS (70 eV): *m/z* 151 (100) [M]⁺, 120 (42), 119 (90), 92 (36), 91 (33), 65 (25).

(*E*)-3-(1*H*-Pyrrol-2-yl)acrylonitrile (5d) and (*Z*)-3-(1*H*-Pyrrol-2-yl)acrylonitrile (5d') [21]: Following the method of preparation for 5b, by using 5a (0.500 g, 5.26 mmol), NaH (60%) (0.252 g, 6.31 mmol), diethylcyanomethylphosphonate (1.117 g, 6.31 mmol) in dry THF (5 mL) and stirring at 25 °C for 24 h, 5d (0.563 g, 91%) and 5d' (0.029 g, 5%) were obtained as colorless liquids.

Data for **5d**: $R_f = 0.56$ (hexane/EtOAc, 7:3). IR (film): $\overline{v} = 3313$, 2924, 2209, 1613, 1445, 1411, 1314, 1128, 1096, 1036, 956, 796, 738 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 5.44$ (d, J = 16.5 Hz, 1H, H-2), 6.22-6.30 (m, 1H, H-4'), 6.50-6.56 (m, 1H, H-3'), 6.91-6.97 (m, 1H, H-5'), 7.20 (d, J = 16.5 Hz, 1H, H-3), 9.46 (br s, 1H, NH). ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 87.0$ (C-2), 110.9 (C-4'), 114.9 (C-3'), 119.9 (C-1), 123.7 (C-5'), 127.7 (C-2'), 140.0 (C-3). MS (70 eV): m/z 118 (100) [M]⁺, 117 (12), 92 (11), 91 (34), 67 (13). Data for **5d**': $R_f = 0.61$ (hexane/EtOAc, 7:3). IR (film): $\overline{v} = 3383, 2924, 2205, 1606, 1447, 1375, 1127, 1094, 1038, 741$ cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 4.96$ (d, J = 11.7 Hz, 1H, H-2), 6.26-6.32 (m, 1H, H-4'), 6.57-6.64 (m, 1H, H-3'), 6.95 (d, J = 11.7 Hz, 1H, H-3), 6.98.-7.04 (m, 1H, H-5'), 9.63 (br s, 1H, NH). ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 84.4$ (C-2), 110.3 (C-4'), 118.0 (C-3'), 120.0 (C-1), 123.7 (C-5'), 127.9 (C-2'), 138.0 (C-3). MS (70 eV): m/z 118 (32) [M]⁺, 111 (88), 109 (56), 97 (84), 95 (93), 85 (49), 83 (59), 81 (100), 71 (58), 69 (69).

Ethyl (E)-3-(1-(Prop-2-yn-1-yl)-1H-pyrrol-2-yl)acrylate (7a): To a solution of 5b (0.200 g, 1.21 mmol) in dry DMF (2.0 mL) at 0 °C and under N2, NaH (60%) (0.058 g, 1.46 mmol) was added. The mixture was stirred at 0 °C for 15 min, and propargyl bromide (6a) (0.144 g, 1.21 mmol) was added dropwise. After stirring at 0 °C for 1 h, EtOAc (20 mL) was added and the mixture was washed with water (2 x 10 mL). The organic layer was dried (Na_2SO_4) and the solvent removed under vacuum. The residue was purified by column chromatography over silica gel (30 g/g crude, hexane/EtOAc, 9:1) to give 7a (0.234 g, 95%) was obtained as a pale yellow oil. $R_f = 0.26$ (hexane/EtO-Ac, 9:1). IR (film): $\overline{v} = 3288$, 1696, 1622, 1468, 1365, 1279, 1265, 1174, 1034, 965, 723 cm⁻¹. ¹H NMR (500 MHz, CDCl₃): $\delta = 1.32$ (t, J = 7.5 Hz, 3H, CO₂CH₂CH₃), 2.46 (t, J = 2.5 Hz, 1H, H-3"), 4.24 (q, J = 7.5 Hz, 2H, CO₂CH₂CH₃), 4.76 (d, J = 2.5 Hz, 2H, H-1"), 6.18 (d, J = 15.5 Hz, 1H, H-2), 6.22 (ddd, J = 4.0, 3.0, 1.0 Hz, 1H, H-4'), 6.68 (dd, J = 4.0, 1.5 Hz, 1.5 Hz)1H, H-3'), 6.94 (dd, J = 3.0, 1.5 Hz, 1H, H-5'), 7.63 (d, J = 15.5Hz, 1H, H-3). ¹³C NMR (125 MHz, CDCl₃): $\delta = 14.4$ (CO₂CH₂CH₂), 36.7 (C-1"), 60.2 (CO₂CH₂CH₂), 74.4 (C-3"), 77.3 (C-2"), 110.0 (C-4'), 112.5 (C-3'), 113.7 (C-2), 125.5 (C-5'), 128.8 (C-2'), 131.6 (C-3), 167.6 (CO₂Et). MS (70 eV): *m*/*z* 203 (8) [M]⁺, 174 (16), 158 (26), 131 (34), 130 (100), 119 (16), 103 (19), 91 (24), 77 (42), 63 (24). HRMS (EI): m/z calcd. for C₁₂H₁₃NO₂ [M]⁺: 203.0946; found 203.0936.

Ethyl (E)-3-(1-(3-Methylbut-2-en-1-yl)-1H-pyrrol-2-yl)acrylate (7b): Following the method of preparation for 7a, by using 5b (0.100 g, 0.61 mmol), NaH (60%) (0.029 g, 0.73 mmol) and prenyl bromide (6b) (0.108 g, 0.73 mmol) in dry DMF (1.0 mL), and after stirring at 0 °C for 1 h, 7b (0.121 g, 85%) was obtained as a reddish oil. $R_{\rm f} = 0.36$ (hexane/EtOAc, 7:3). IR (film): $\overline{v} = 2977$, 1700, 1622, 1472, 1444, 1324, 1276, 1163, 1037, 968, 723 cm⁻¹. ¹H NMR (500 MHz, CDCl₂): $\delta =$ 1.31 (t, J = 7.0 Hz, 3H, CO₂CH₂CH₃), 1.75 (d, J = 1.0 Hz, 3H, H-4" or H-5"), 1.79 (br s, 3H, H-5" or H-4"), 4.23 (q, J = 7.0 Hz, 2H, $CO_2CH_2CH_3$), 4.57 (d, J = 7.0 Hz, 2H, H-1"), 5.28 (tm, J = 6.8 Hz, 1H, H-2"), 6.13 (d, J = 16.0 Hz, 1H, H-2),6.16-6.17 (ddd, J = 3.5, 2.5, 0.5 Hz, 1H, H-4'), 6.64 (dd, J = 3.5, 2.5, 0.5 Hz, 100 Hz, 100 Hz, 100 Hz3.5, 1.5 Hz, 1H, H-3'), 6.79 (dd, J=2.5, 1.5 Hz, 1H, H-5'), 7.60 (d, J = 16.0 Hz, 1H, H-3). ¹³C NMR (125 MHz, CDCl₂): $\delta =$ 14.4 (CO₂CH₂CH₃), 17.9 (C-4" or C-5"), 25.6 (C-5" or C-4"), 45.2 (C-1"), 60.1 (CO₂CH₂CH₃), 109.3 (C-4'), 111.5 (C-3'), 112.7 (C-2), 120.0 (C-2"), 125.3 (C-5'), 128.7 (C-2'), 132.4 (C-3), 136.5 (C-3"), 167.8 (CO₂Et). MS (70 eV): *m/z* 233 (12) $[M]^+$, 208 (25), 194 (70), 184 (57), 168 (71), 154 (100), 144 (61), 130 (82), 117 (91), 115 (65), 91 (44), 77 (26). HRMS (EI): m/z calcd. for C₁₄H₁₉NO₂ [M]⁺: 233.1416; found 233.1413.

Ethyl (E)-3-(1-(Propa-1,2-dien-1-yl)-1H-pyrrol-2-yl)acrylate (7c): Following the method of preparation for 7a, by using 5b (0.100 g, 0.61 mmol), NaH (60%) (0.048 g, 1.21 mmol) and propargyl bromide (6a) (0.144 g, 1.21 mmol) in dry DMF (1.0 mL), and after stirring at 0 °C for 1 h, 7c (0.111 g, 90%) was obtained as a colorless oil. $R_f = 0.23$ (hexane/EtOAc, 9:1). IR (film): $\overline{v} = 2925$, 1702, 1623, 1463, 1258, 1175, 1037, 852, 722 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 1.32$ (t, J = 7.2 Hz, 3H, $CO_2CH_2CH_3$), 4.24 (q, J = 7.2 Hz, 2H, $CO_2CH_2CH_3$), 5.53 (d, J = 6.5 Hz, 2H, H-3"), 6.19 (d, J = 15.5 Hz, 1H, H-2), 6.27 (dd, J = 3.9, 2.7 Hz, 1H, H-4'), 6.69 (dd, J = 3.9, 1.5 Hz, 1H, H-3'), 6.94 (dd, J = 2.7, 1.5 Hz, 1H, H-5'), 7.04 (t, J = 6.5 Hz, 1H, H-1"), 7.69 (d, J = 15.5 Hz, 1H, H-3). ¹³C NMR (75.4 MHz, $CDCl_3$): $\delta = 14.3 (CO_2CH_2CH_3), 60.3 (CO_2CH_2CH_3), 86.9 (C-$ 3"), 97.0 (C-1"), 111.1 (C-4'), 112.5 (C-3'), 114.1 (C-2), 123.9 (C-5'), 128.9 (C-2'), 131.7 (C-3), 167.5 (CO₂Et), 203.8 (C-2"). MS (70 eV): *m/z* 203 (43) [M]⁺, 175 (100), 158 (30), 130 (71), 129 (47), 103 (21), 91 (10), 77 (18). HRMS (EI): m/z calcd. for C₁₂H₁₃NO₂ [M]⁺: 203.0946; found 203.0945.

Ethyl (E)-3-(1-(2-Ethoxy-2-oxoethyl)-1H-pyrrol-2-yl)acrylate (7d): Following the method of preparation for 7a, by using 5b (0.050 g, 0.30 mmol), NaH (60%) (0.015 g, 0.38 mmol) and ethyl bromoacetate (6c) (0.061 g, 0.36 mmol) in dry DMF (1.0 mL), and after stirring at 0 °C for 1 h, 7d (0.075 g, 99%) was obtained as a colorless oil. $R_f = 0.49$ (hexane/EtOAc, 7:3). IR (film): $\overline{v} = 2982$, 1751, 1699, 1471, 1366, 1326, 1290, 1263, 1207, 1175, 967, 728 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 1.30 (q, J = 7.1 Hz, 6H, 2 CO_2 CH_2 CH_3), 4.23 (q, J = 7.1 Hz, 6H, 2 CO_2 CH_2 CH_3)$ 2H, CO₂CH₂CH₃), 4.24 (q, J = 7.1 Hz, 2H, CO₂CH₂CH₃), 4.74 (s, 2H, H-1"), 6.16 (d, J = 15.3 Hz, 1H, H-2), 6.25 (ddd, J = 3.6)2.4, 0.9 Hz, 1H, H-4'), 6.72 (dd, J = 3.6, 1.5 Hz, 1H, H-3'), 6.79 (dd, *J* = 1.5, 0.9 Hz, 1H, H-5'), 7.45 (d, *J* = 15.3 Hz, 1H, H-3). ¹³C NMR (75.4 MHz, CDCl₂): $\delta = 14.1$ (CO₂CH₂CH₂), 14.3 (CO₂CH₂CH₃), 48.4 (C-1"), 60.2 (CO₂CH₂CH₃), 62.0 (CO₂CH₂CH₃), 110.3 (C-4'), 112.2 (C-3'), 113.8 (C-2), 126.7 (C-5'), 129.4 (C-2'), 131.5 (C-3), 167.6 (CO₂Et), 168.0 (CO₂Et). MS (70 eV): *m/z* 251 (74) [M]⁺, 223 (5), 206 (44), 179 (32), 133 (31), 104 (100), 103 (94), 102 (61), 78 (26). HRMS (EI): m/z calcd. for $C_{13}H_{17}NO_4 [M]^+$: 251.1158; found 251.1158.

Ethyl (*E*)-3-(1-(Furan-2-carbonyl)-1*H*-pyrrol-2-yl)acrylate (7e): Following the method of preparation for 7a, by using 5b (0.050 g, 0.30 mmol), NaH (60%) (0.015 g, 0.38 mmol) and 2-furoyl chloride (6d) (0.040 g, 0.36 mmol) in dry DMF (1.0 mL), and after stirring at 0 °C for 2.5 h, 7e (0.069 g, 87%) was obtained as a reddish oil. $R_f = 0.43$ (hexane/EtOAc, 7:3). IR (film): $\bar{v} = 1698$, 1623, 1567, 1465, 1391, 1339, 1271, 1178, 1031, 853, 765 cm⁻¹. ¹H NMR (500 MHz, CDCl₃): $\delta = 1.31$ (t, J = 7.0 Hz, 3H, CO₂CH₂CH₃), 4.23 (q, J = 7.0 Hz, 2H, CO₂CH₂CH₃), 6.25 (d, J = 16.0 Hz, 1H, H-2), 6.34 (dd, J = 3.5, 3.0 Hz, 1H, H-4'), 6.65 (dd, J = 3.5, 1.5 Hz, 1H, H-4''), 6.83 (dm, J = 3.5 Hz, 1H, H-3'), 7.36 (dd, J = 3.5, 0.5 Hz, 1H, H-3''), 7.50 (dd, J = 3.0, 1.5 Hz, 1H, H-5'), 7.72 (dd, J = 1.5, 0.5 Hz, 1H, H-5"), 8.05 (d, J = 16.0 Hz, 1H, H-3). ¹³C NMR (125 MHz, CDCl₃): $\delta = 14.3$ (CO₂CH₂CH₃), 60.3 (CO₂CH₂CH₃), 112.4 (C-4'), 112.6 (C-4"), 115.5 (C-3'), 116.9 (C-2), 122.0 (C-3"), 125.4 (C-5'), 132.1 (C-2'), 134.0 (C-3), 146.4 (C-2"), 147.6 (C-5"), 156.9 (NCO), 166.9 (CO₂Et). MS (70 eV): m/z 259 (50) [M]⁺, 231 (11), 203 (32), 186 (18), 158 (22), 119 (12), 95 (100), 63 (9). HRMS (EI): m/z calcd. for C₁₄H₁₃NO₄ [M]⁺: 259.0845; found 259.0851.

Methyl (E)-3-(1-(Prop-2-yn-1-yl)-1H-pyrrol-2-yl)acrylate (8a): Following the method of preparation for 7a, by using 5c (0.200 g, 1.33 mmol), NaH (60%) (0.064 g, 1.59 mmol) and propargyl bromide (6a) (0.189 g, 1.59 mmol) in dry DMF (2.0 mL), and after stirring at 0 °C for 1.5 h, 8a (0.233 g, 93%) was obtained as a colorless oil. $R_{\rm f} = 0.70$ (hexane/EtOAc, 7:3). IR (film): $\overline{v} = 3243, 2951, 1690, 1622, 1468, 1331, 1286, 1173,$ 972, 731 cm⁻¹. ¹H NMR (300 MHz, CDCl₂): $\delta = 2.46$ (t, J = 2.4Hz, 1H, H-3"), 3.78 (s, 3H, CO_2CH_3), 4.75 (d, J = 2.4 Hz, 2H, H-1"), 6.18 (d, J = 15.6 Hz, 1H, H-2), 6.20-6.23 (m, 1H, H-4'), 6.69 (dd, J = 3.9, 1.5 Hz, 1H, H-3'), 6.94 (dd, J = 2.7, 1.5 Hz, 1H, H-5'), 7.64 (d, J = 15.6 Hz, 1H, H-3). ¹³C NMR (75.4 MHz, $CDCl_3$): $\delta = 36.7 (C-1'')$, $51.5 (CO_2CH_3)$, 74.5 (C-3''), 77.3 (C-2"), 110.0 (C-4'), 112.6 (C-3'), 113.2 (C-2), 125.6 (C-5'), 128.6 (C-2'), 131.8 (C-3), 168.0 (CO₂Me). MS (70 eV): *m*/*z* 189 (17) [M⁺], 174 (40), 158 (43), 130 (100), 118 (11), 103 (22), 91 (14), 77 (19). Anal. calcd. for C₁₁H₁₁NO₂: C, 69.83; H, 5.86; N, 7.40. Found: C, 69.80; H, 5.82; N, 7.40.

Methyl (E)-3-(1-(Propa-1,2-dien-1-yl)-1H-pyrrol-2-yl)acrylate (8b): Following the method of preparation for 7a, by using 5c (0.100 g, 0.66 mmol), NaH (60%) (0.053 g, 1.33 mmol) and propargyl bromide (6a) (0.156 g, 1.33 mmol) in dry DMF (1.0 mL), and after stirring at 0 °C for 1 h, 8b (0.107 g, 85%) was obtained as a dark solid. $R_f = 0.77$ (hexane/EtOAc, 7:3); mp 54-55 °C. IR (KBr): $\overline{v} = 2924$, 1704, 1622, 1460, 1436, 1263, 1168, 1030, 966, 723 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta =$ 3.77 (s, 3H, CO₂CH₃), 5.51 (d, J = 6.3 Hz, 1H, H-3"), 6.19 (d, J = 15.6 Hz, 1H, H-2), 6.26 (dd, J = 3.9, 3.0 Hz, 1H, H-4'), 6.69 (dd, J = 3.9, 1.5 Hz, 1H, H-3'), 6.93 (dd, J = 3.0, 1.5 Hz, 1H, H-5'), 7.02 (t, J = 6.3 Hz, 1H, H-1''), 7.69 (d, J = 15.6 Hz, 1H, H-3). ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 51.4$ (CO₂CH₃), 86.8 (C-3"), 96.8 (C-1"), 111.1 (C-4'), 112.5 (C-3'), 113.5 (C-2), 124.0 (C-5'), 128.8 (C-2'), 131.9 (C-3), 167.8 (CO₂CH₃), 203.8 (C-2"). MS (70 eV): *m/z* 189 (11) [M]⁺, 158.0 (7), 130 (100), 103 (20), 77 (13). Anal. calcd. for C₁₁H₁₁NO₂ (189.21): C, 69.83; H, 5.86; N, 7.40. Found: C, 69.81; H, 5.86; N, 7.39.

(*E*)-3-(1-(Prop-2-yn-1-yl)-1*H*-pyrrol-2-yl)acrylonitrile (9a): Following the method of preparation for 7a, by using 5d (0.200 g, 1.70 mmol), NaH (60%) (0.081 g, 2.03 mmol) and propargyl bromide (6a) (0.242 g, 2.03 mmol) in dry DMF (2.0 mL), and after stirring at 0 °C for 1.5 h, 9a (0.229 g, 87%) was obtained as a colorless oil. $R_{\rm f} = 0.66$ (hexane/EtOAc, 7:3). IR (KBr): $\overline{v} = 3237, 2208, 1608, 1470, 1412, 1295, 1085, 951, 728 {\rm cm}^{-1}.$ ¹H NMR (300 MHz, CDCl₃): $\delta = 2.50$ (t, J = 2.4 Hz, 1H, H-3"), 4.71 (d, J = 2.4 Hz, 2H, H-1"), 5.56 (d, J = 16.2 Hz, 1H, H-2), 6.22 (ddd, J = 3.7, 2.6, 0.6 Hz, 1H, H-4'), 6.68 (dd, J = 3.7, 1.5 Hz, 1H, H-3'), 6.93 (dd, J = 2.6, 1.5 Hz, 1H, H-5'), 7.32 (d, J = 16.2 Hz, 1H, H-3). ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 36.8$ (C-1"), 74.9 (C-3"), 76.9 (C-2"), 90.6 (C-2), 110.3 (C-4'), 112.6 (C-3'), 119.3 (C-1), 126.6 (C-5'), 128.0 (C-2'), 137.0 (C-3). MS (70 eV): m/z 156 (40) [M]⁺, 155 (100), 130 (18), 128 (24). Anal. calcd. for C₁₀H₈N₂: C, 76.90; H, 5.16; N, 17.94. Found: C, 76.86; H, 5.13; N, 17.93.

(E)-3-(1-(Propa-1,2-dien-1-yl)-1H-pyrrol-2-yl)acrylonitrile (9b): Following the method of preparation for 7a, by using 5d (0.100 g, 0.85 mmol), NaH (60%) (0.068 g, 1.70 mmol) and propargyl bromide (6a) (0.200 g, 1.70 mmol) in dry DMF (1.0 mL), and after stirring at 0 °C for 1 h, 9b (0.114 g, 86%) was obtained as a colorless oil. $R_{\rm f} = 0.74$ (hexane/EtOAc, 7:3). IR (film): $\overline{v} = 2925, 2209, 1610, 1463, 1410, 1320, 1134, 956, 892,$ 722 cm⁻¹. ¹H NMR (300 MHz, CDCl₂): $\delta = 5.55$ (d, J = 6.3, 1H, H-3"), 5.56 (d, J = 16.2 Hz, 1H, H-2), 6.27 (dd, J = 3.8, 2.7 Hz, 1H, H-4'), 6.67 (dd, J = 3.8, 1.5 Hz, 1H, H-3'), 6.91 (t, J = 6.3 Hz, 1H, H-1"), 6.95 (dd, J = 2.7, 1.5 Hz, 1H, H-5'), 7.34 (d, J = 16.2 Hz, 1H, H-3). ¹³C NMR (75.4 MHz, CDCl₂): $\delta = 87.0$ (C-3"), 90.7 (C-2), 96.4 (C-1"), 111.2 (C-4'), 112.5 (C-3'), 119.0 (C-1), 124.8 (C-5'), 128.0 (C-2'), 137.0 (C-3), 203.7 (C-2"). MS (70 eV): *m/z* 156 (100) [M]⁺, 155 (100), 130 (20), 128 (42), 101 (6).

Ethyl (E)-3-(5-Formyl-1-(prop-2-ynyl)-1H-pyrrol-2-yl)acrylate (10a): Dry DMF (0.043 g, 0.59 mmol) was added to phosphorus oxychloride (0.091 g, 0.59 mmol) at 0 °C, and the resulting mixture was stirred for 10 min. Then, 7a (0.100 g, 0.49 mmol) was added dropwise and the temperature was slowly raised to 40 °C and maintained for 2 h. The reaction mixture was quenched with an aqueous solution of NaOH 2N until neutral, CH₂Cl₂ (250 mL) was added, the organic layer dried (Na_2SO_4) and the solvent removed under vacuum. The residue was purified by column chromatography over silica gel (30 g/g crude, hexane/EtOAc, 98:2) to give 10a (0.101 g, 88%) as a white solid. $R_f = 0.59$ (hexane/EtOAc, 7:3); mp 90-91 °C. IR (KBr): $\overline{v} = 1705$, 1659, 1628, 1472, 1449, 1409, 1309, 1270, 1234, 1183, 1048, 783 cm^{'1}. ¹H NMR (300 MHz, CDCl₃): $\delta = 1.35$ (t, J = 7.2 Hz, 3H, CO₂CH₂CH₃), 2.38 (t, J = 2.7 Hz, 1H, H-3"), 4.28 (q, J = 7.2 Hz, 2H, CO₂CH₂CH₃), 5.39 (d, J = 2.7 Hz, 2H, H-1"), 6.47 (d, J = 15.6 Hz, 1H, H-2), 6.70 (dd, J = 4.2, 0.6 Hz, 1H, H-3'), 6.96 (dd, J = 4.2, 0.9 Hz, 1H, H-4'), 7.72 (d, J = 15.6 Hz, 1H, H-3), 9.60 (s, 1H, CHO). ¹³C NMR $(75.4 \text{ MHz}, \text{CDCl}_3): \delta = 14.3 (\text{CO}_2\text{CH}_2\text{CH}_3), 34.4 (\text{C}-1''), 60.8$ (CO₂CH₂CH₃), 73.6 (C-3"), 77.7 (C-2"), 111.6 (C-3'), 121.4 (C-2), 124.7 (C-4'), 130.0 (C-3), 132.8 (C-5'), 137.5 (C-2'), 166.3 (CO₂Et), 179.9 (CHO). MS (70 eV): *m/z* 231 (M⁺, 18), 203 (41), 202 (100), 174 (37), 158 (41), 130 (42), 103 (24), 77 (9). Anal. calcd. for C₁₃H₁₃NO₃: C, 67.52; H, 5.67; N, 6.06. Found: C, 67.52; H, 5.61; N, 6.03.

Ethyl (*E*)-3-(5-Formyl-1-(3-methylbut-2-enyl)-1*H*-pyrrol-2-yl)acrylate (10b): Following the method of preparation for 10a, by using DMF (0.038 g, 0.52 mmol), POCl₃ (0.080 g, 0.52 mmol) and 7b (0.100 g, 0.43 mmol), and after stirring at 40 °C for 2 h, 10b (0.091 g, 81%) was obtained as a white solid. $R_{\rm f} = 0.34$ (hexane/EtOAc, 7:3); mp 74-75 °C. IR (film): $\overline{v} = 1706, 1663, 1625, 1475, 1307, 1244, 1184, 1051, 772 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): $\delta = 1.34$ (t, J = 6.9 Hz, 3H, CO₂CH₂CH₃), 1.71 (br s, 3H, H-4" or H-5"), 1.85 (br s, 3H, H-5" or H-4"), 4.26 (q, J = 6.9 Hz, 2H, CO₂CH₂CH₃), 5.15 (br s, 3H, H-1", H-2"), 6.39 (d, J = 15.9 Hz, 1H, H-2), 6.66 (dd, $J = 4.2, 0.6 \text{ Hz}, 1\text{H}, \text{H}-3^{\circ}), 6.92 \text{ (dd}, J = 4.2, 0.6 \text{ Hz}, 1\text{H}, \text{H}-4^{\circ}),$ 7.60 (d. J = 15.9 Hz. 1H. H-3), 9.58 (s. 1H. CHO), ¹³C NMR $(75.4 \text{ MHz}, \text{CDCl}_3)$: $\delta = 14.3 (\text{CO}_2\text{CH}_2\text{CH}_3)$, 18.2 (C-4" or C-5"), 25.6 (C-5" or C-4"), 43.4 (C-1"), 60.7 (CO₂CH₂CH₃), 110.9 (C-3'), 120.3 (C-2), 120.4 (C-2"), 124.3 (C-4'), 130.8 (C-3), 133.3 (C-5'), 135.5 (C-3''), 137.3 (C-2'), 166.5 (CO₂Et), 179.8 (CHO). MS (70 eV): m/z 261 (M⁺, 7), 225 (15), 202 (17), 188 (56), 167 (20), 155 (20), 148 (100), 119 (24), 73 (20). HRMS (EI): m/z calcd. for C₁₅H₁₉NO₃ [M]⁺: 261.1365; found 261.1358.

Ethyl (E)-3-(5-Formyl-1-(propa-1,2-dien-1-yl)-1H-pyrrol-2-yl)acrylate (10c): Following the method of preparation for **10a**, by using DMF (0.086 g, 1.18 mmol), POCl₂ (0.181 g, 1.18 mmol) and 7c (0.200 g, 0.99 mmol), and after stirring at 40 °C for 2 h, 10c (0.189 g, 83%) was obtained as a pale yellow oil. $R_{\rm f} = 0.34$ (hexane/EtOAc, 7:3). IR (film): $\overline{v} = 1709$, 1660, 1445, 1370, 1261, 1190, 1036, 784 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 1.34$ (t, J = 7.1 Hz, 3H, CO₂CH₂CH₂), 4.26 (q, J = 7.1 Hz, 2H, $CO_2CH_2CH_3$), 5.48 (d, J = 6.4 Hz, 2H, H-3"), 6.37 (d, J = 15.8 Hz, 1H, H-2), 6.70 (dd, J = 4.3, 0.6 Hz, 1H, H-3'), 7.02 (dd, J = 4.3, 0.6 Hz, 1H, H-4'), 7.62 (t, J = 6.4 Hz, 1H, H-1''), 7.83 (dd, J = 15.8, 0.6 Hz, 1H, H-3), 9.65 (s, 1H, CHO). ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 14.2$ (CO₂CH₂CH₃), 60.7 (CO₂CH₂CH₃), 85.6 (C-3"), 95.9 (C-1"), 111.8 (C-3"), 120.2 (C-2), 124.0 (C-4'), 131.8 (C-3), 133.4 (C-5'), 137.3 (C-2'), 166.4 (CO₂Et), 179.5 (CHO), 203.8 (C-2"). MS (70 eV): m/z 231 (46) [M]⁺, 203 (12), 158 (100), 130 (53), 103 (27), 77 (9). HRMS (EI): m/z calcd. for C₁₃H₁₃NO₃ [M]⁺: 231.0895; found 231.0882.

Methyl (E)-3-(5-Formyl-1-(propa-1,2-dien-1-yl)-1H-pyrrol-2-yl)acrylate (11a): Following the method of preparation for 10a, by using DMF (0.047 g, 0.64 mmol), POCl₃ (0.098 g, 0.64 mmol) and 8b (0.100 g, 0.53 mmol), and after stirring at 40 °C for 2 h, 11a (0.101 g, 88%) was obtained as a brown solid. $R_{\rm f} = 0.65$ (hexane/EtOAc, 7:3); mp 88-89 °C. IR (film): $\bar{v} = 2950$, 1706, 1656, 1627, 1449, 1310, 1275, 1201, 1046, 971, 790 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 3.81$ (s, 3H, CO₂CH₃), 5.48 (d, J = 6.6 Hz, 2H, H-3"), 6.37 (d, J = 15.6 Hz, 1H, H-2), 6.70 (d, J = 4.3 Hz, 1H, H-3'), 7.02 (d, J = 4.3 Hz, 1H, H-4'), 7.60 $(t, J = 6.6 \text{ Hz}, 1\text{H}, \text{H}-1^{"}), 7.82 \text{ (d}, J = 15.6 \text{ Hz}, 1\text{H}, \text{H}-3), 9.65$ (s, 1H, CHO). ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 51.8$ (CO₂CH₃), 85.5 (C-3"), 95.9 (C-1"), 111.8 (C-3'), 119.6 (C-2), 123.9 (C-4'), 132.0 (C-3), 133.5 (C-5'), 137.1 (C-2'), 166.8 (CO₂CH₃), 179.4 (CHO), 203.6 (C-2"). MS (70 eV): m/z 217 (42) [M]⁺, 158 (100), 130 (37), 103 (22), 77 (11). Anal. calcd. for C₁₂H₁₁NO₃: C, 66.35; H, 5.10; N, 6.45. Found: 66.30; H, 5.12; N, 6.45.

(E)-3-(5-Formyl-1-(prop-2-ynyl)-1H-pyrrol-2-yl)acrylonitrile (12a): Following the method of preparation for 10a, by using DMF (0.056 g, 0.77 mmol), POCl₃ (0.118 g, 0.77 mmol) and 9a (0.100 g, 0.64 mmol), and after stirring at 40 °C for 2 h, 12a (0.109 g, 92%) was obtained as a white solid. $R_f = 0.51$ (hexane/EtOAc, 7:3); mp 103-104 °C. IR (film): $\bar{v} = 2213$, 1662, 1614, 1452, 1412, 1236, 1213, 1051, 968, 786 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 2.43$ (t, J = 2.7 Hz, 1H, H-3"), 5.38 (d, J = 2.7 Hz, 2H, H-1"), 5.91 (d, J = 16.2 Hz, 1H, H-2), 6.71 (dd, J = 4.4, 0.6 Hz, 1H, H-3'), 6.97 (dd, J = 4.4, 0.6 Hz, 1H, H-4'), 7.47 (br d, J = 16.2 Hz, 1H, H-3), 9.63 (s, 1H, CHO). ¹³C NMR (75.4 MHz, CDCl₃): δ = 34.4 (C-1"), 74.2 (C-3"), 77.3 (C-2"), 98.6 (C-2), 111.7 (C-3'), 117.7 (C-1), 124.5 (C-4'), 133.1 (C-5'), 136.1 (C-3), 136.2 (C-2'), 180.3 (CHO). MS (70 eV): *m/z* 184 (36) [M]⁺, 156 (34), 155 (100), 138 (34), 128 (29). HRMS (EI): *m/z* calcd. for C₁₁H₈N₂O [M]⁺: 184.0637; found 184.0637.

(E)-3-(5-Formyl-1-(propa-1,2-dien-1-yl)-1H-pyrrol-2-yl)acrylonitrile (12b): Following the method of preparation for 10a, by using DMF (0.112 g, 1.54 mmol), POCl₃ (0.236 g, 1.54 mmol) and 9b (0.200 g, 1.28 mmol), and after stirring at 40 °C for 2 h, 12b (0.175 g, 75%) was obtained as a colorless oil. $R_f = 0.32$ (hexane/EtOAc, 7:3). IR (film): $\bar{v} = 2922, 2207,$ 1607, 1458, 1412, 1303, 1079, 951, 731 cm⁻¹. ¹H NMR (300 MHz, CDCl₂): $\delta = 5.51$ (d, J = 6.6 Hz, 2H, H-3"), 5.81 (d, J = 16.5 Hz, 1H, H-2), 6.70 (br d, J = 4.2 Hz, 1H, H-3'), 7.03 (dd, J = 4.2, 0.6 Hz, 1H, H-4'), 7.50 (dt, J = 16.5, 0.6 Hz, 1H)H-3), 7.59 (t, J = 6.6 Hz, 1H, H-1"), 9.68 (s, 1H, CHO). ¹³C NMR (75.4 MHz, CDCl₃): δ = 85.9 (C-3"), 95.7 (C-1"), 97.4 (C-2), 111.9 (C-3'), 117.8 (C-1), 123.7 (C-4'), 133.8 (C-5'), 135.8 (C-2'), 137.4 (C-3), 179.7 (CHO), 206.3 (C-2"). MS (70 eV): m/z 184 (70) $[M]^+$, 156 (100), 128 (30). HRMS (EI): m/z calcd. for C₁₁H₂N₂O [M]⁺: 184.0637; found 184.0635.

4-(Hydroxymethyl)phenyl 4-Methylbenzenesulfonate (18): Triethylamine (0.829 g, 8.19 mmol) was added to a mixture of 4-hydroxybenzaldehyde (17) (0.500 g, 4.09 mmol) and DMAP (0.050 g, 0.41 mmol) in CH₂Cl₂ (15 mL), which was stirred at room temperature for 20 min. p-Toluenesulfonyl chloride (1.171 g, 6.14 mmol) was added at 0 °C and the mixture was stirred for 1.5 h. The solvent was removed under vacuum, the residue was dissolved in MeOH (10 mL) and NaBH₄ (0.080 g, 2.05 mmol) was added at 0 °C. The reaction mixture was stirred at the same temperature for 2 h. The solvent was removed under vacuum and the residue purified by column chromatography over silica gel (10 g/g crude, hexane/EtOAc, 1:1) to give 18 (1.187 g, 77%) as a pale yellow oil. $R_f = 0.17$ (hexane/EtOAc, 7:3). IR (film): v = 3368, 2925, 1597, 1502, 1369, 1197, 1174, 1150, 1092, 1015, 865, 814, 691 cm⁻¹. ¹H NMR (500 MHz, $CDCl_3$): $\delta = 2.20$ (br, 1H, OH), 2.44 (s, 3H, CH_3), 4.63 (s, 2H, CH_2), 6.94 (d, J = 8.8 Hz, 2H, H-2'), 7.25 (d, J = 8.8 Hz, 2H, H-3'), 7.29 (d, J = 8.0 Hz, 2H, H-3), 7.68 (d, J = 8.0 Hz, 2H, H-2). ¹³C NMR (125 MHz, CDCl₃): δ = 21.6 (*C*H₃), 64.2 (*C*H₂), 122.3 (C-2'), 127.9 (C-3'), 128.4 (C-2), 129.7 (C-3), 132.2 (C-1), 139.8 (C-4'), 145.4 (C-4), 148.8 (C-1'). MS (70 eV): *m/z* 278 (40) [M⁺], 155 (54), 139 (10), 123 (31), 107 (46), 91 (100), 77 (10). HRMS (EI): *m/z* calcd. for C₁₄H₁₄O₄S [M]⁺: 278.0613; found 278.0604.

4-(Bromomethyl)phenyl 4-Methylbenzenesulfonate (19): Triphenylphosphine (0.860 g, 3.80 mmol) was added to a solution of 18 (0.760 g, 2.73 mmol) in CH₂Cl₂ (20 mL). After stirring at room temperature for 10 min, NBS (0.580 g, 3.28 mmol) was added at 0 °C, and the mixture was stirred at this temperature for 1 h. The solvent was removed under vacuum and the residue purified by column chromatography over silica gel (10 g/g crude, hexane/EtOAc, 95:5) to afford 19 (0.876 g, 94%) as a white solid. $R_f = 0.63$ (hexane/EtOAc, 7:3); mp 78-79 °C. IR (KBr): $\overline{v} = 3401, 2918, 1806, 1596, 1500, 1369, 1197, 1177,$ 1147, 1091, 1018, 859, 810, 742, 695 cm⁻¹. ¹H NMR (500 MHz, $CDCl_3$): $\delta = 2.44$ (s, 3H, CH_3), 4.42 (s, 2H, CH_2), 6.95 (d, J = 8.5 Hz, 2H, H-2'), 7.28-7.33 (m, 4H, H-3', H-3), 7.70 (d, J = 8.0 Hz, 2H, H-2). ¹³C NMR (125 MHz, CDCl₂): $\delta = 21.7$ (CH₂), 32.1 (CH₂), 122.7 (C-2'), 128.4 (C-2), 129.8 (C-3), 130.3 (C-3'), 132.3 (C-1), 136.7 (C-4'), 145.5 (C-4), 149.3 (C-1'). MS (70 eV): *m/z* 341 (2) [M⁺], 261 (81), 155 (100), 91 (77). HRMS (EI): *m/z* calcd. for C₁₄H₁₃BrO₃S [M]⁺: 339.9769; found 339.9759.

4-((2-Formyl-1*H*-pyrrol-1-yl)methyl)phenyl 4-Methylbenzenesulfonate (20): According to the method for the preparation of 7a, by using 2-formylpyrrol (5a) (0.558 g, 5.86 mmol), NaH (60%) (0.281 g, 7.03 mmol) and compound 19 (2.00 g, 5.86 mmol) in dry DMF (10 mL), and after stirring at 25 °C for 5 h, **20** (2.001 g, 96%) was obtained as a white resin. $R_f = 0.38$ (hexane/EtOAc, 7:3). IR (film): $\overline{v} = 1662, 1597, 1503, 1477,$ 1404, 1371, 1319, 1198, 1177, 1154, 1092, 867, 757, 720 cm⁻¹. ¹H NMR (500 MHz, CDCl₃): $\delta = 2.43$ (s, 3H, CH₃), 5.50 (s, 2H, CH_2), 6.27 (t, J = 3.6 Hz, 1H, H-4"), 6.86-6.92 (m, 2H, H-2'), 6.96 (d, J = 3.6 Hz, 2H, H-3", H-5"), 7.01-7.07 (m, 2H, H-3'), 7.25-7.32 (m, 2H, H-3), 7.63-7.69 (m, 2H, H-2), 9.50 (d, J=0.6 Hz, 1H, CHO). ¹³C NMR (125 MHz, CDCl₃): $\delta = 21.6$ (CH₃), 51.0 (CH₂), 110.3 (C-4"), 122.4 (C-2'), 125.0 (C-3"), 128.2 (C-3'), 128.3 (C-2), 129.7 (C-3), 131.1 (C-2"), 131.4 (C-5"), 132.0 (C-1), 136.6 (C-4'), 145.4 (C-4), 148.8 (C-1'), 179.4 (CHO). MS (70 eV): *m/z* 355 (27) [M]⁺, 337 (8), 261 (14), 200 (94), 155 (54), 107 (32), 94 (100), 91 (38).

4-((2-(Hydroxymethyl)-1*H***-pyrrol-1-yl)methyl)phenyl 4-Methylbenzenesulfonate (21): NaBH₄ (0.099 g, 2.62 mmol) was added to a solution of 20** (1.860 g, 5.23 mmol) in a mixture of MeOH/CH₂Cl₂ (1:1) (20 mL) at 0 °C, and the mixture was stirred at this temperature for 2 h. The solvent was removed under vacuum and the crude product purified by column chromatography over silica gel (20 g/g crude, hexane/EtOAc, 7:3) to afford **21** (1.777 g, 95%) as a colorless oil. R_f = 0.25 (hexane/EtOAc, 7:3). IR (film): \bar{v} = 3375, 2924, 1596, 1501, 1367, 1298, 1196, 1174, 1150, 1091, 1016, 862, 715 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 1.51 (br s, 1H, OH), 2.44 (s, 3H, *CH*₃), 4.45 (br s, 2H, *CH*₂OH), 5.15 (s, 2H, H-1"), 6.11 (dd, *J* = 3.5, 2.7 Hz, 1H, H-4""), 6.15 (dd, *J* = 3.5, 1.8 Hz, 1H, H-3""), 6.65 (dd, *J* = 2.7, 1.8 Hz, 1H, H-5""), 6.86-6.97 (m, 4H, H-2', H-3'), 7.26-7.33 (m, 2H, H-3), 7.65-7.71 (m, 2H, H-2). ¹³C NMR (75.4 MHz, CDCl₃): δ = 21.7 (*C*H₃), 49.7 (C-1"), 56.6 (*C*H₂OH), 107.5 (C-4""), 109.5 (C-3""), 122.6 (C-2"), 123.1 (C-5""), 127.7 (C-3'), 128.4 (C-2), 129.7 (C-3), 131.6 (C-2""), 132.1 (C-1), 137.3 (C-4'), 145.4 (C-4), 148.7 (C-1'). HRMS (EI): *m/z* calcd. for C₁₉H₁₉NO₄S [M]⁺: 357.1035; found 357.1025.

(1-(4-(*p*-Tosyloxy)benzyl)-1*H*-pyrrol-2-yl)methyl acetate (22): Pyridine (0.806 g, 10.19 mmol) was added to a solution of 21 (1.821 g, 5.09 mmol) in dry CH₂Cl₂ (30 mL) at 0 °C. After stirring for 30 min, acetic anhydride (1.048 g, 10.19 mmol) was added and the mixture was stirred at room temperature for 24 h, followed by washing with water (100 mL) and an aqueous solution of HCl 5% until neutral. The aqueous layer was extracted with CH₂Cl₂ (2 x 100 mL). The combined organic layers were dried (Na_2SO_4) and the solvent was removed under vacuum. The residue was purified by column chromatography over silica gel (20 g/g crude, hexane/EtOAc, 8:2) to give 22 (1.974 g, 97%) as a yellow oil. $R_f = 0.48$ (hexane/EtOAc, 7:3). IR (film): $\overline{v} = 2926, 1735, 1596, 1502, 1372, 1302, 1237, 1197, 1176,$ 1152, 1092, 1017, 864, 834, 815, 719 cm⁻¹. ¹H NMR (300 MHz, $CDCl_3$): $\delta = 1.80$ (s, 3H, CH_3CO_2), 2.45 (s, 3H, CH_3), 4.97 (s, 2H, CH₂), 5.09 (s, 2H, H-1"), 6.16 (dd, J = 3.5, 2.8 Hz, 1H, H-4'), 6.29 (dd, J=3.5, 1.8 Hz, 1H, H-3'), 6.70 (dd, J=2.8, 1.8 Hz, 1H, H-5'), 6.85-6.95 (m, 4H, H-2"', H-3"'), 7.28-7.34 (m, 2H, H-3^{IV}), 7.66-7.72 (m, 2H, H-2^{IV}). ¹³C NMR (75.4 MHz, $CDCl_3$): $\delta = 20.7 (CH_3CO_2), 21.7 (CH_3), 49.8 (C-1"), 57.4$ (CH₂), 108.0 (C-4'), 112.0 (C-3'), 122.6 (C-3'''), 123.7 (C-5'), 126.6 (C-2'), 127.3 (C-2"'), 128.4 (C-2^{IV}), 129.8 (C-3^{IV}), 132.2 (C-1^{IV}), 137.2 (C-1^{IV}), 145.4 (C-4^{IV}), 148.7 (C-4^{IV}), 170.6 (CH₃CO₂). MS (70 eV): *m/z* 341 (14) [M - 59]⁺, 261 (13), 203 (17), 184 (15), 155 (22), 124 (20), 107 (24), 91 (100), 77 (25). HRMS (EI): m/z calcd. for C₂₁H₂₁NO₅S [M]⁺: 399.1140; found 399.1129.

(5-Formyl-1-(4-(p-tosyloxy)benzyl)-1H-pyrrol-2-yl)methyl acetate (23): Following the method of preparation for 7a, by using dry DMF (0.137 g, 1.87 mmol), POCl₃ (0.287 g, 1.87 mmol) and 22 (0.680 g, 1.70 mmol) in dry DMF (6.0 mL), and after stirring at 0 °C for 30 min, the reaction mixture was quenched with an aqueous solution of KOH 1M (30 mL) and extracted with EtOAc (2 x 100 mL). The organic layer was dried (Na_2SO_4) and the solvent was removed under vacuum. The residue was purified by column chromatography over silica gel (20 g/g crude, hexane/EtOAc, 8:2) to give 23 (0.209 g, 57%) as a pale yellow resin. $R_f = 0.63$ (hexane/EtOAc, 1:1). IR (film): $\overline{v} = 1741$, 1663, 1503, 1371, 1225, 1198, 1176, 1153, 1092, 1019, 865 cm⁻¹. ¹H NMR (500 MHz, CDCl₃): $\delta = 1.84$ (s, 3H, CH₃CO₂), 2.45 (s, 3H, CH₃), 5.00 (s, 2H, CH₂), 5.64 (s, 2H, H-1"), 6.49 (d, J = 4.0 Hz, 1H, H-3"), 6.83-6.88 (m, 2H, H-3""), 6.88-6.93 (m, 2H, H-2"), 6.97 (d, J = 4.0 Hz, 1H, H-4'), 7.28-7.32 (m, 2H, H-3^{IV}), 7.67-7.98 (m, 2H, H-2^{IV}), 9.55 (s, 1H, CHO). ¹³C NMR (125 MHz, CDCl₃): $\delta = 20.5$ (CH₃CO₂), 21.7 (CH₃), 48.0 (C-1"), 56.7 (CH₂), 112.9 (C-3"), 122.5 (C-3""), 124.2 (C-4"), 127.2 (C-2""), 128.4 (C-2^{IV}), 129.8 (C-3^{IV}), 132.4 (C-1^{IV}), 132.8 (C-5'), 136.6 (C-1""), 137.0 (C-2"); 145.4 (C-4^{IV}), 148.8 (C-4""), 170.2 (CHO), 179.8 (CH₃CO₂). MS (70 eV): *m/z* 427 (3) [M]⁺, 367 (11), 272 (9), 261 (9), 212 (100), 196 (17), 184 (24), 155 (52), 123 (38), 108 (11), 91 (27). HRMS (EI): *m/z* calcd. for C₂₂H₂₁NO₆S [M]⁺: 427.1090; found 427.1089.

1-(4-Hydroxybenzyl)-5-(hydroxymethyl)-1H-pyrrole-2-carbaldehyde (Pyrrolemarumine) (4b) [10]: A mixture of compound 23 (0.167 g, 0.39 mmol) and KOH (0.088 g, 1.56 mmol) in a mixture of MeOH/H₂O (1:1) (3 mL) was stirred at room temperature for 24 h. MeOH was removed under vacuum and CH₂Cl₂ (50 mL) was added. The mixture was washed with water (50 mL) and an aqueous solution of HCl 5% until neutral. The aqueous layer was extracted with CH₂Cl₂ (2 x 100 mL). The organic layers were dried (Na2SO4) and the solvent removed under vacuum. The residue was purified by column chromatography over silica gel (20 g/g crude, hexane/EtOAc, 1:1) to afford **4b** (0.069 g, 76%) as a pale yellow resin. $R_f = 0.35$ (hexane/EtOAc, 1:1). IR (film): $\overline{v} = 3333$, 2970, 1644, 1516, 1456, 1371, 1243, 1173, 1046, 1011, 822, 784, cm⁻¹. ¹H NMR $(300 \text{ MHz}, \text{ acetone-} d_6): \delta = 4.48 \text{ (br s, 1H, CH}_2\text{O}H), 4.57 \text{ (br s, })$ 2H, CH₂OH), 5.65 (s, 2H, H-1'), 6.27 (d, *J* = 3.9 Hz, 1H, H-4), 6.71-6.78 (m, 2H, H-3"), 6.89-6.96 (m, 2H, H-2"), 6.98 (d, J = 3.9 Hz, 1H, H-3), 8.43 (br s, 1H, Ar-OH), 6.53 (s, 1H, CHO). ¹³C NMR (75.4 MHz, acetone- d_6): $\delta = 48.3$ (C-1'), 56.7 (CH₂OH), 110.7 (C-4), 116.1 (C-3"), 124.8 (C-3), 128.7 (C-2"), 130.0 (C-1"), 133.3 (C-2), 144.4 (C-5), 157.5 (C-4"), 180.0 (CHO). MS (70 eV): *m/z* 231 (11) [M]⁺, 214 (11), 125 (100), 108 (82), 107 (80), 96 (48), 77 (39), 68 (18). DART-MS m/z calcd. for $C_{13}H_{14}NO_3[M + H]^+$: 232.0967; found: 232.0974.

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